

(19) World Intellectual Property Organization
International Bureau(43) International Publication Date
6 June 2002 (06.06.2002)

PCT

(10) International Publication Number
WO 02/44430 A1(51) International Patent Classification⁷: C21D 1/613,
9/00, 1/767Francis [GB/GB]; 42 Park Lea, Bradley, Huddersfield,
West Yorkshire HD2 1QH (GB).

(21) International Application Number: PCT/GB01/05308

(74) Agent: WICKHAM, Michael; The Boc Group Plc, Chertsey Road, Windlesham, Surrey GU20 6HJ (GB).

(22) International Filing Date:
30 November 2001 (30.11.2001)(81) Designated States (national): AU, CA, CN, CZ, JP, KR,
RU, US, ZA.

(25) Filing Language: English

(84) Designated States (regional): Eurasian patent (AM, AZ,
BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE,
CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC,
NL, PT, SE, TR).(30) Priority Data:
0029281.3 30 November 2000 (30.11.2000) GB**Published:**

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

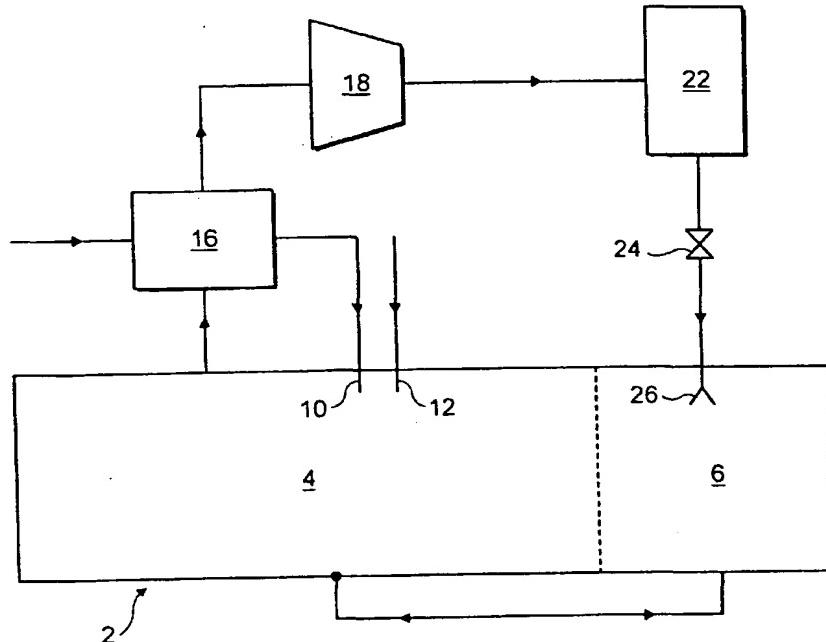
(71) Applicant (for all designated States except US): THE BOC GROUP PLC [GB/GB]; Chertsey Road, Windlesham, Surrey GU20 6HJ (GB).

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(72) Inventor; and

(75) Inventor/Applicant (for US only): STRATTON, Paul,

(54) Title: QUENCHING METHOD AND APPARATUS



WO 02/44430 A1

(57) Abstract: A hot metal object is quenched after heat treatment. A hot gas stream comprising at least 20 % by volume of hydrogen is taken from for example a carburising chamber (4) of a furnace (2). The gas is cooled by passage through a heat exchanger (16), and is compressed in a compressor (18). The compressor (18) has an aftercooler (not shown) to remove heat of compression from the gas. The cooled, compressed gas flows through nozzles (26) into a quenching chamber (6). The gas leaves the nozzles (26) at a velocity of at least 50 m/s and impinges upon the hot metal object so as to effect its quenching.

QUENCHING METHOD AND APPARATUS

This invention relates to a method of and apparatus for quenching a hot metal object.

It is very well known in the art of heat treating metal that quenching a metallic object (that is rapidly cooling the object from a heat treatment temperature, typically at least 850°C, to a much lower, usually room, temperature) can significantly improve its mechanical properties and characteristics. For example, quenching can be used to harden the object and/or to improve its mechanical properties, by controlling internal crystallisation or precipitation, or both. Traditionally, quenching has been carried out using liquid such as water, oil or brine, either in the form of an immersion bath or a spraying medium. In more recent years, gas quenching methods have been developed. Gas quenching has the advantage of not usually requiring an after quenching step to clean or wash the quenched metal object. Another advantage of gas quenching is that if an oil or water-based fluid is used non-uniformity problems can arise as a result of Leidenfrost's phenomenon, whereas in gas quenching, this problem is believed not to arise.

The main drawback of the gas quenching which has to now limited its commercial use is a difficulty in achieving a quench rate comparable to those that characterise liquid-based cooling methods. Gas quenching is discussed in "Innovations in Quenching Systems and Equipment: Current Status and Feature Developments", by FT Hoffmann et al, Heat Treatment of Metals, 1999, 3, pp 63 to 67. Hoffmann et al do not however disclose how to form a hydrogen quenching atmosphere.

Gas quenching is also disclosed in EP-A-0 911 418 and in US-A-5 770 146. GB-A-1 394 197 describes the operation of a furnace for annealing coiled steel strip. The furnace has a series of five cooling sections which employ recycled gas from

-2-

the annealing section. The recycled gas is coded and supplied to the cooling sections by means of jet nozzles. A Roots-type blower may be used to recirculate the gas from the annealing section to the nozzles. Cooling rates of up to 25°C per hour are achieved. Such cooling rates are to be contrasted with the high cooling rates of at least 50°C per hour that characterise gas quenching.

According to the present invention there is provided a method of quenching a hot metal object by taking a hot gas stream comprising at least 20% by volume of hydrogen from a source thereof, cooling the hot gas stream, compressing the cooled gas stream removing heat of compression from the cool pressed gas stream, passing the compressed gas through at least one nozzle and causing the gas issuing from the said nozzle to impinge upon the hot metal object so as to quench the object, wherein the source of the hot gas is a heat treatment chamber from which the hot metal objects taken for quenching or a gas generator which supplies hot gas to the heat treatment chamber.

The invention also provides apparatus for quenching a hot metal object taken from a heat treatment chamber, comprising a source of hot gas containing at least 20% by volume of hydrogen, a heat exchanger for cooling the hot gas having an inlet communicating with the source and an outlet communicating with an inlet to a compressor an aftercooler associated with the compressor, a quenching chamber, means for introducing the hot metal object into the quenching chamber, at least one nozzle arranged so as to cause, in use gas to impinge upon the object to be quenched in the quenching chamber, the said nozzle communicating with an outlet from the compressor, wherein the source of the hot gas is the heat treatment chamber or a gas generator which is able to supply hot gas containing at least 20% by volume of hydrogen to the heat treatment chamber.

By employing the heat treatment chamber or gas generator as the source of the quenching gas, the need for a separate supply of hydrogen to the quenching step is obviated.

Although the method and apparatus according to the present invention may be employed in annealing the metal object, they are particularly suitable if the metal object is to be hardened, carburised, case hardened or carbonitrited and are able to treat effectively metal objects of complex shapes.

The hot gas is typically taken from the heat treatment chamber or the generator at a temperature in the range of 850°C to 950°C. If the heat treatment for example, comprises carburising the metal object, the hot gas preferably contains from 25 to 40% by volume of hydrogen. The hot gas may in addition contain from 40 to 60% by volume of nitrogen, from 12 to 20% by volume of carbon monoxide, with smaller amounts of other gases such as methane, water vapour, and carbon dioxide typically also being present. If the heat treatment comprises carbonitriding or austenitic nitrocarburising the metal object the atmosphere may also include ammonia.

The stream of hot gas is preferably compressed to a pressure up to 10 bar gauge, the maximum pressure not being so great that the dew point of the gas is less than 15°C, thus ensures that water does not precipitate out of the gas stream.

A carburising gas stream may be formed in an endothermic generator or, preferably, by supplying nitrogen and a precursor of both carbon monoxide and hydrogen to the carburising chamber and permitting the precursor to decompose in the carburising chamber to form carbon monoxide and hydrogen. The preferred precursor is methanol. One advantage of forming the carburising gas in such a way rather than in an endothermic generator is that it enables the composition of the atmosphere to

be adjusted by adjusting the flow rates of the nitrogen and the precursor. For example, if the precursor is methanol, its flow rate can be selected so as to give the minimum water content in the resulting gaseous atmosphere in the carburising chamber, and thereby maximising the pressure to which the gas stream withdrawn from the carburising chamber can be compressed. Preferably, the atmosphere is formed by supplying to the carburising chamber 55 volumes of nitrogen to every 45 volumes of methanol.

The heat treatment chamber is preferably operated at a pressure in the range of 0 bar gauge to 1 bar gauge.

The hot gas stream taken from the heat treatment chamber is preferably cooled by indirect heat exchange with a stream of nitrogen. If the nitrogen is to be supplied to the treatment chamber, this has the added advantage of preheating the nitrogen. The cooled gas stream preferably leaves the heat exchanger at a temperature less than 50°C.

Preferably, a gas storage vessel is located intermediate the compressor outlet and the said nozzle. Such an arrangement keeps down the power consumption of the method and apparatus according to the invention when the quenching is performed intermittently.

Typically, depending on the size of the object to be cooled, a plurality of nozzles is used in the method and apparatus according to the invention. Preferably, the distance between each nozzle outlet and the surface at which the gas issuing from the nozzle is directed is less than or equal to the diameter of the nozzle. Such a distance is selected in view of our discovery that at small values of the distance between the nozzle outlet and the surface of the object there is a surprisingly large increase in the heat transfer rate as the distance decreases.

-5-

Preferably the distance between adjacent nozzle outlets is in the range of from 2 to 8 times the diameter of each nozzle.

Preferably the or each nozzle directs gas so as to impinge substantially perpendicularly on the surface of the object.

Because the rate of cooling during quenching is directly related to the velocity of the gas streams, and the velocity to the gas supply pressure, it is a relatively simple matter to control the cooling rate. The preferred gas velocities are at least 50 metres per second, more preferably in the range of 50 to 100 metres per second. Typical nozzle diameters are in the range of 3.2 to 6.4 mm.

Preferably, there is a conduit having one end terminating in the quenching chamber and another end terminating in the heat treatment chamber. This allows spent gas from the quenching chamber to flow to the heat treatment chamber. The conduit also enables reducing gas to be supplied to the quenching chamber when quenching is not taking place provided that the pressure in the heat treatment chamber is maintained slightly above that in the quenching chamber when the latter is idle.

The heat treatment chamber and the quenching chamber may form part of the same furnace, for example a roller hearth furnace. If the furnace has a cooling chamber intermediate the heat treatment chamber and the quenching chamber, the reducing gas may be withdrawn from the cooling chamber. This, however, is not preferred as the dew point of the atmosphere is greater in the cooling chamber.

The method according to the invention will now be described by way of example with reference to the accompanying drawing which is a schematic flow diagram of a roller hearth furnace which has been adapted to perform the invention.

Referring to the drawing, a roller hearth furnace 2 has a carburising chamber 4 and a quenching chamber 6. The furnace also includes a belt (not shown) for transporting work to be carburised into the furnace 2, through the carburising chamber 4, then through the quenching chamber 6 and out of the furnace 2. The carburising chamber 4 has a first inlet 10 for nitrogen and a second inlet 12 for methanol. The positioning of the inlets may be conventional. The furnace is provided with a heater (not shown) so as to raise the temperature of the atmosphere in the carburising chamber 4 to a temperature in the range 850 to 950°C. Under these conditions, the methanol, if supplied in liquid form, will evaporate. Gaseous methanol cracks at the temperatures prevailing in the carburising chamber 4 to form hydrogen and carbon monoxide. Preferably, for each 55 moles of nitrogen, 45 moles of methanol are supplied to the carburising chamber 4. As a result, an atmosphere containing approximately 55% by volume of nitrogen, 30% by volume of hydrogen, and 15% by volume of carbon monoxide is formed, excluding minor impurities such as methane, water vapour and carbon dioxide. Typically, the water vapour content of this atmosphere is only to about 0.26%. A stream of the atmosphere is withdrawn from the carburising chamber 4 and passes through a heat exchanger 16 in which it is cooled to a temperature in the order of 50°C by heat exchange with ambient temperature nitrogen upstream of the introduction of the nitrogen into the chamber 4 through the inlet 10. As a result, the nitrogen is preheated and this reduces the amount of thermal energy that needs to be supplied to the carburising chamber 4 by the internal heater (not shown).

The resulting cooled gas stream is compressed to a pressure of 7 bar g (8 bar absolute) in a compressor 18. The compressor 18 is preferably operated

continuously and is sized such that the flow rate therethrough is less than that required for quenching. The compressor 18 is provided with an aftercooler (not shown) so as to remove heat of compression from the compressed gas. The compressed gas is supplied to a pressure vessel 22 in which it is stored. The pressure vessel 22 has a valved outlet 24 communicating with an array of nozzles 26 for directing gas at the object to be quenched in the quenching chamber 6. For ease of illustration, only one of the nozzles 26 is shown in the drawing. The distance from the nozzle outlet to the surface of the metal object against which the gas impinges is in the range of from a quarter to a half the nozzle diameter. Typically, the nozzle has a diameter in the range of 6.4 to 12.8 mm.

The actual flow rate of gas from the pressure vessel 22 to the nozzles 26 is greater than the rate at which gas flows into the pressure vessel 22. The normal operation of the furnace 2 is, however, such that the quenching chamber 6 is used only intermittently. Thus, the pressure vessel 22 can be so operated that it always contains a supply of quenching gas at pressure. While the quenching chamber 6 receives gas from the nozzles 26, the spent gas passes via a conduit 30 back into the carburising chamber 4. On the other hand during periods when the quenching chamber 6 is not being used, gas is able to pass from the carburising chamber 4 into it via the conduit 30 so as to maintain reducing conditions therein.

In view of the hydrogen content of the quenching gas, a quenching rate may be achieved in the chamber which can equal or exceed that achieved by conventional medium quench oils. Such a rapid quenching rate is achieved without the disadvantages attendant upon use of quenching oils, namely the need to clean the work after it has been quenched and the risk of some structural distortion being created by the quenching oil.

CLAIMS

1. A method of quenching a hot metal object by taking a hot gas stream comprising at least 20% by volume of hydrogen from a source thereof, cooling the hot gas stream, compressing the cooled gas stream, removing heat of compression from the compressed gas stream, passing the compressed gas through at least one nozzle and causing the gas issuing from the said nozzle to impinge upon the hot metal object so as to quench the object, wherein the source of the hot gas is a heat treatment chamber from which the hot metal object is taken for quenching or a gas generator which supplies hot gas to the heat treatment chamber.
2. A method as claimed in claim 1, in which the heat treatment is annealing, hardening, carburising, case hardening, carbonitriding, or austenitic carbonitriding.
3. A method as claimed in claim 1 or claim 2, in which the hot gas is taken from the heat treatment chamber or generator at a temperature in the range of 850°C to 950°C.
4. A method as claimed in any one of the preceding claims, in which the heat treatment comprises characterising the metal object and the hot gas contains from 25 to 40% by volume of hydrogen.
5. A method as claimed in claim 4, in which the hot gas additionally contains from 40 to 60% by volume of nitrogen and from 12 to 20% by volume of carbon monoxide.

-9-

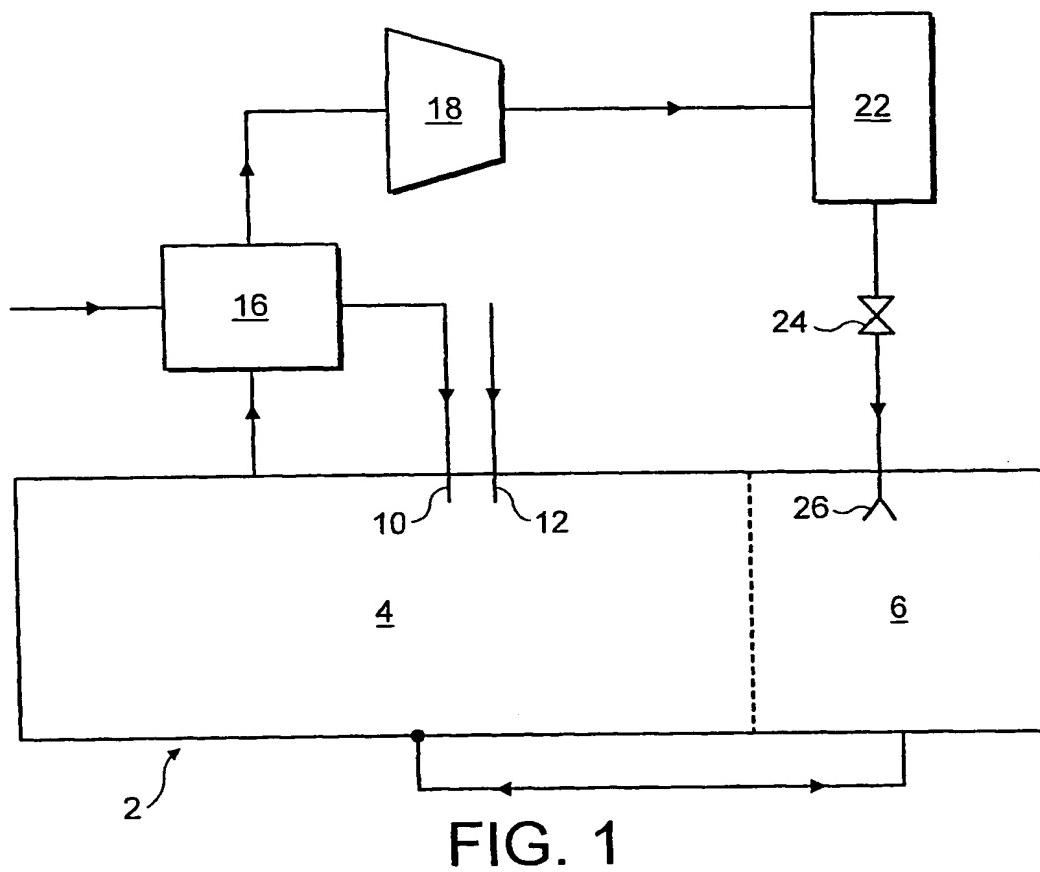
6. A method as claimed in any one of the preceding claims, in which the stream of hot gas is compressed to a pressure up to 10 bar gauge and the compressed gas has a dew point less than 15°C.
7. A method as claimed in any one of the preceding claims, in which the heat treatment chamber is operated at a pressure in the range of 0 bar gauge to 1 bar gauge.
8. A method as claimed in any one of the preceding claims, in which a gas storage vessel is located intermediate the compressor outlet and the said nozzle.
9. A method as claimed in any one of the preceding claims, in which the distance between the said nozzle outlet and the surface at which the gas issuing from the nozzle is directed is less than or equal to the diameter of the nozzle.
10. A method as claimed in any one of the preceding claims, in which a gas issues from the said nozzle at a velocity of at least 50 metres per second.
11. Apparatus for quenching a hot metal object taken from of a heat treatment chamber, comprising a source of hot gas containing at least 20% by volume of hydrogen, a heat exchanger for cooling the hot gas having an inlet communicating with the source and an outlet communicating with an inlet to a compressor, an aftercooler associated with the compressor, a quenching chamber, means for introducing the hot metal object into the quenching chamber and at least one nozzle arranged so as to cause, in use, gas to impinge upon the object to be quenched in the quenching chamber, the said nozzle communicating with an outlet from the compressor, wherein the source of the hot gas is the heat treatment chamber or a gas generator which is able

-10-

to supply hot gas containing at least 20% by volume to the heat treatment chamber.

12. Apparatus as claimed in claim 11, in which the said nozzle communicates with the outlet of the compressor via a pressure vessel.

1 / 1



SUBSTITUTE SHEET (RULE 26)

INTERNATIONAL SEARCH REPORT

Int'l application No

PCT/GB 01/05308

A. CLASSIFICATION OF SUBJECT MATTER
 IPC 7 C21D1/613 C21D9/00 C21D1/767

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
 IPC 7 C21D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 911 418 A (NIPPON STEEL CORP) 28 April 1999 (1999-04-28) cited in the application ---	
A	GB 1 394 197 A (ALLEGHENY LUDLUM IND INC) 14 May 1975 (1975-05-14) cited in the application ---	
A	DE 198 53 221 A (LINDE AG) 25 May 2000 (2000-05-25) ---	
A	DE 42 08 485 C (J. WÜNNING) 11 February 1993 (1993-02-11) ---	
A	WO 89 12111 A (WINGENS ULRICH) 14 December 1989 (1989-12-14) ---	

 Further documents are listed in the continuation of box C. Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

Date of mailing of the International search report

11 April 2002

17/04/2002

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
 NL - 2280 HV Rijswijk
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
 Fax: (+31-70) 340-3016

Authorized officer

Mollet, G

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No.

PCT/GB 01/05308

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
EP 0911418	A	28-04-1999		BR 9804782 A EP 0911418 A1 CN 1219206 A CN 1219206 T WO 9841661 A1 TW 404982 B	17-08-1999 28-04-1999 09-06-1999 09-06-1999 24-09-1998 11-09-2000
GB 1394197	A	14-05-1975		AT 329104 B AT 787972 A AU 474698 B AU 4634972 A BE 788908 A1 CA 970964 A1 DE 2244913 A1 ES 406750 A1 FR 2152930 A1 GB 1394198 A GB 1394199 A GB 1394200 A IT 965376 B JP 48038214 A JP 56027580 B JP 1145016 C JP 56077325 A JP 57035247 B SE 440367 B SE 440368 B SE 7511643 A US 3778221 A	26-04-1976 15-07-1975 29-07-1976 14-03-1974 15-03-1973 15-07-1975 22-03-1973 01-10-1975 27-04-1973 14-05-1975 14-05-1975 14-05-1975 31-01-1974 05-06-1973 25-06-1981 12-05-1983 25-06-1981 28-07-1982 29-07-1985 29-07-1985 16-10-1975 11-12-1973
DE 19853221	A	25-05-2000	DE	19853221 A1	25-05-2000
DE 4208485	C	11-02-1993	DE	4208485 C1	11-02-1993
			AT	160382 T	15-12-1997
			DE	59307686 D1	02-01-1998
			EP	0562250 A1	29-09-1993
			JP	6010037 A	18-01-1994
			US	5452882 A	26-09-1995
WO 8912111	A	14-12-1989	DE	3819803 C1	14-12-1989
			WO	8912111 A1	14-12-1989